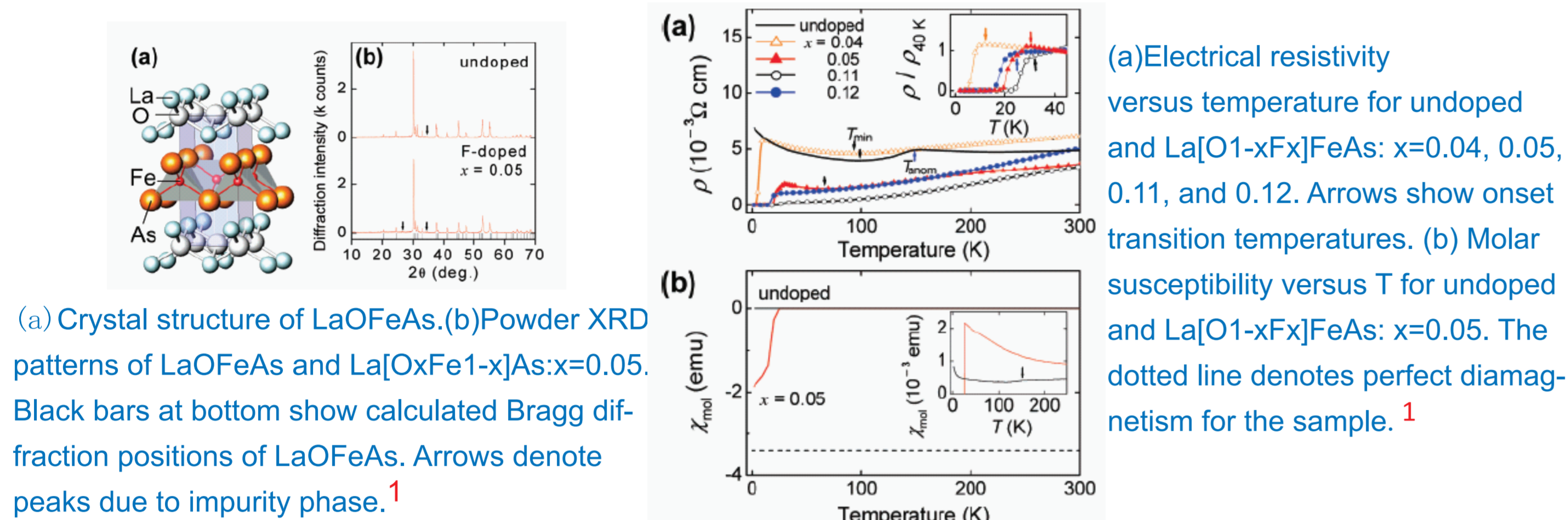


Fei Chen¹, Cheng He¹, Min Xu¹ and D. L. Feng^{1*}

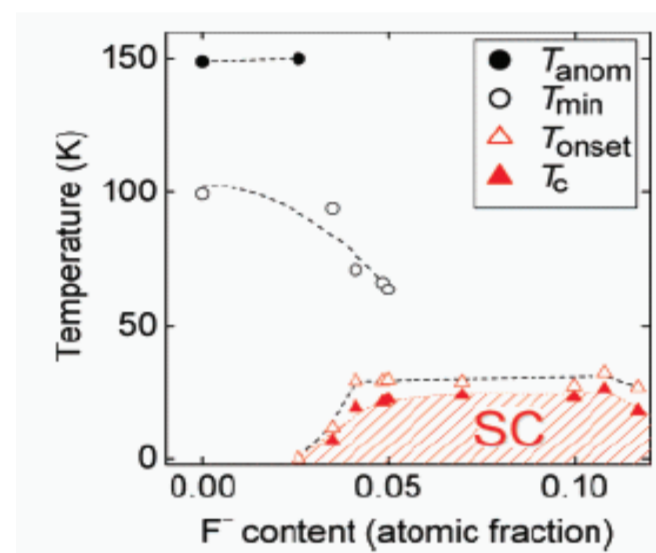
¹Department of Physics, Applied Surface Physics State Key Laboratory, Fudan University, Shanghai 200433, P. R. China

Introduction

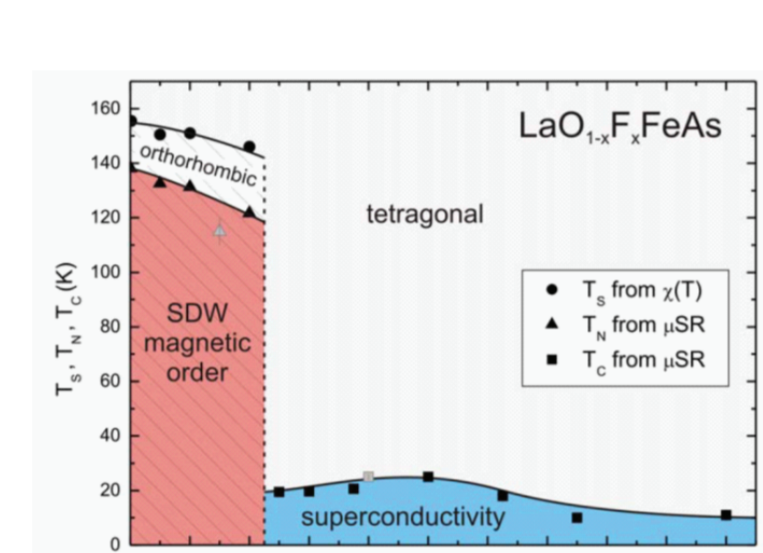
The iron-based material has been a hot topic this year since the “1111” La[O1-xFx]FeAs¹ with T_c=26K firstly discovered by the Japanese Scientists. Soon the other SC Re[O1-xFx]FeAs(Re=rare earth) with T_c up to 56K were made in series. Then the other systems “122” [Ba1-xKx]Fe2As2 and “11” α-FeSe were discovered in succession. Because the iron-based material is the second high temperature superconductor besides the cuprates, our group has also tried to reveal the superconducting mechanism behind this new material. So we began to grow the iron-based material and then tested their structures, compositions and transport properties.



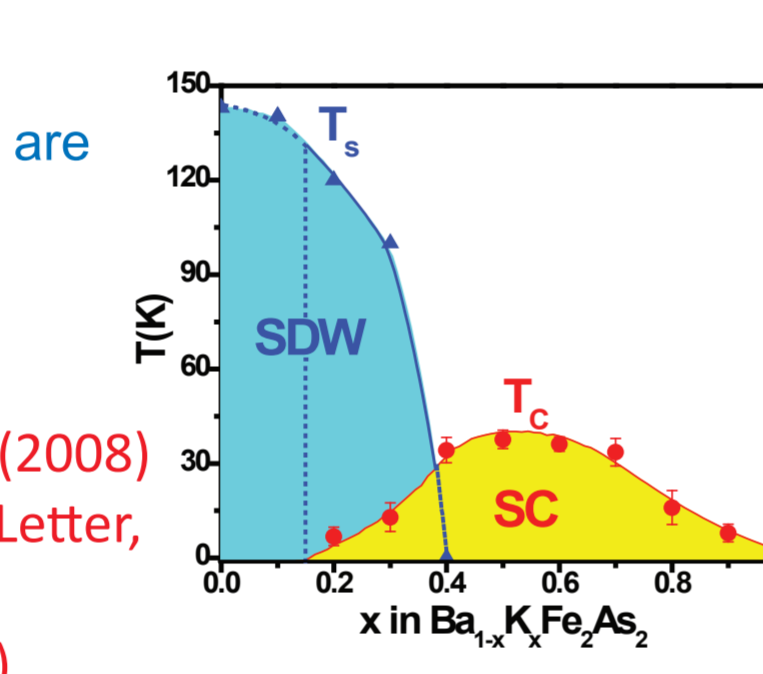
(a) Crystal structure of LaOFeAs. (b) Powder XRD patterns of LaOFeAs and La[O1-xFx]FeAs: x=0.05. Black bars at bottom show calculated Bragg diffraction positions of LaOFeAs. Arrows denote peaks due to impurity phase.¹



T_c, T_{onsset}, and T_{min} in the ρ-T curves as a function of F⁻ content(x) for La[O1-xFx]FeAs. T_c is defined as the temperature where the ρ value becomes half of that at T_{onsset}. Tanom values for the undoped and LaO0.97F0.03FeAs are also shown. Dotted curves are guides for eyes.¹



The doping dependence of the magnetic and superconducting transition temperatures determined from the μSR experiments. Also shown are the tetragonal to orthorhombic structural transition temperatures T_s determined from resistivity measurements which show a kink and subsequent strong reduction below T_s.²

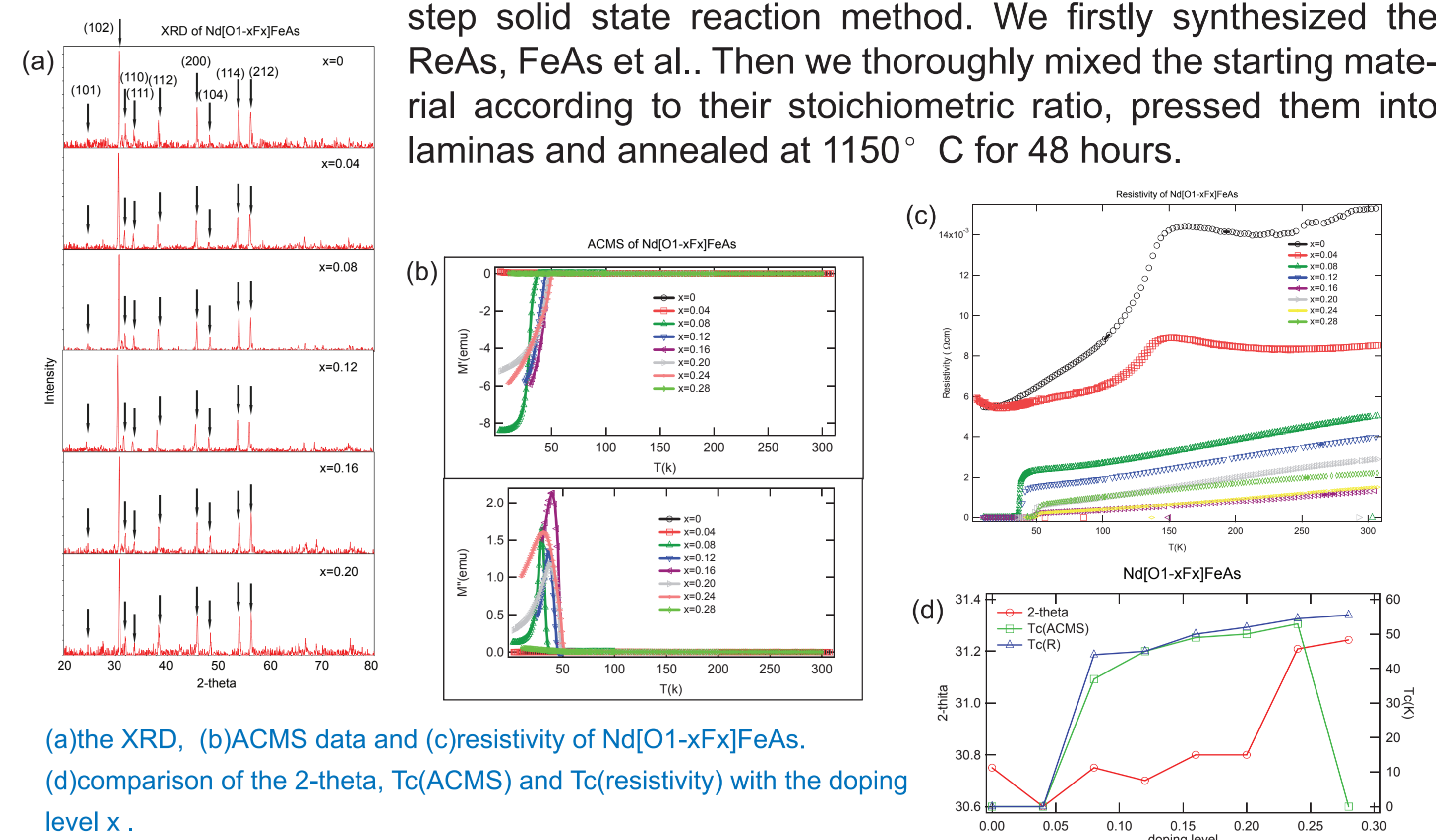


The composition-temperature phase diagram, showing the structure, magnetic and superconducting transitions. The T_s denotes the temperature of the simultaneous structural and magnetic transition, and T_c the superconducting one.³

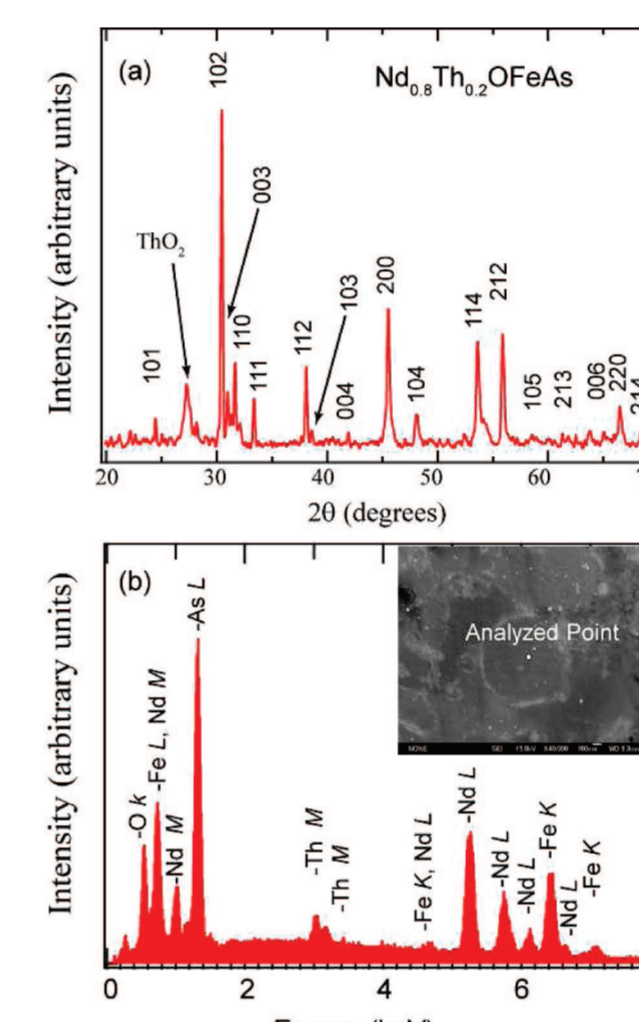
Ref 1. Y. Kamihara et al., J. Am. Chem. Soc. 130, 3296(2008)
Ref 2. H. Luetkens et al., Nature Materials Letter, 10.1038/NMAT2397
Ref 3. H. Chen et al., Europhys. Lett. 85, 17006 (2009)

the Polycrystals of the “1111” system

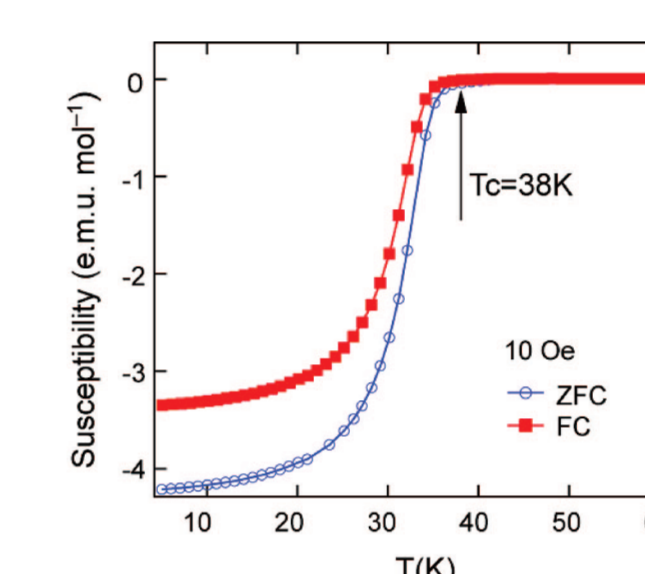
The polycrystals of the “1111” system were grown by the two step solid state reaction method. We firstly synthesized the ReAs, FeAs et al.. Then we thoroughly mixed the starting material according to their stoichiometric ratio, pressed them into laminas and annealed at 1150° C for 48 hours.



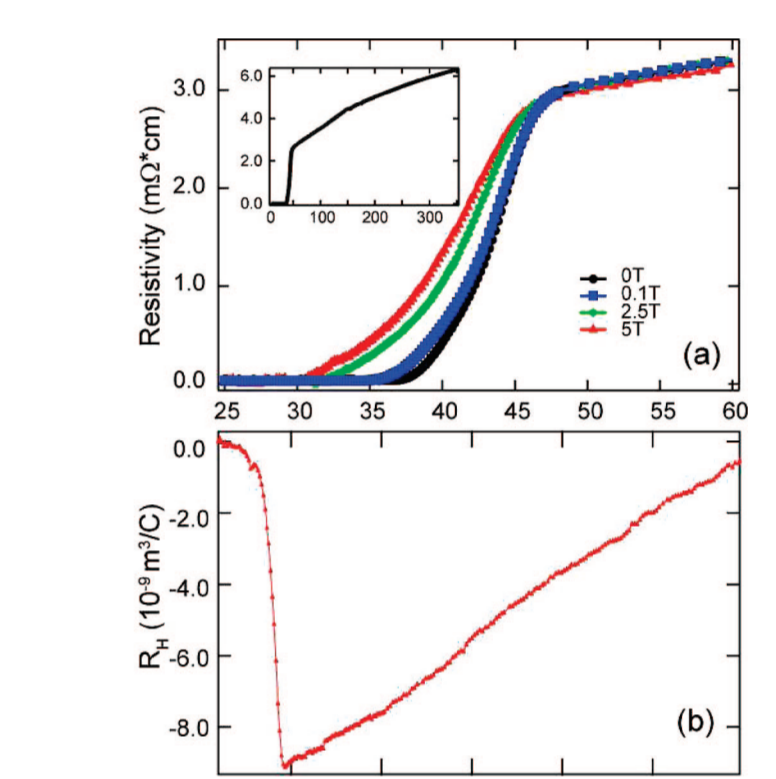
(a) the XRD. (b) ACMS data and (c) resistivity of Nd[O1-xFx]FeAs. (d) comparison of the 2-theta, T_c(ACMS) and T_c(resistivity) with the doping level x.



Powder XRD (a) and EDX data (b) of Nd_{0.8}Th_{0.2}OFeAs. The inset in panel b shows the scanning electron microscope picture. The white spot shows where the EDX data were taken.⁴



Temperature dependence of the magnetic susceptibility of Nd_{0.8}Th_{0.2}OFeAs under 10 Oe in zero-field cooling and field cooling process.⁴

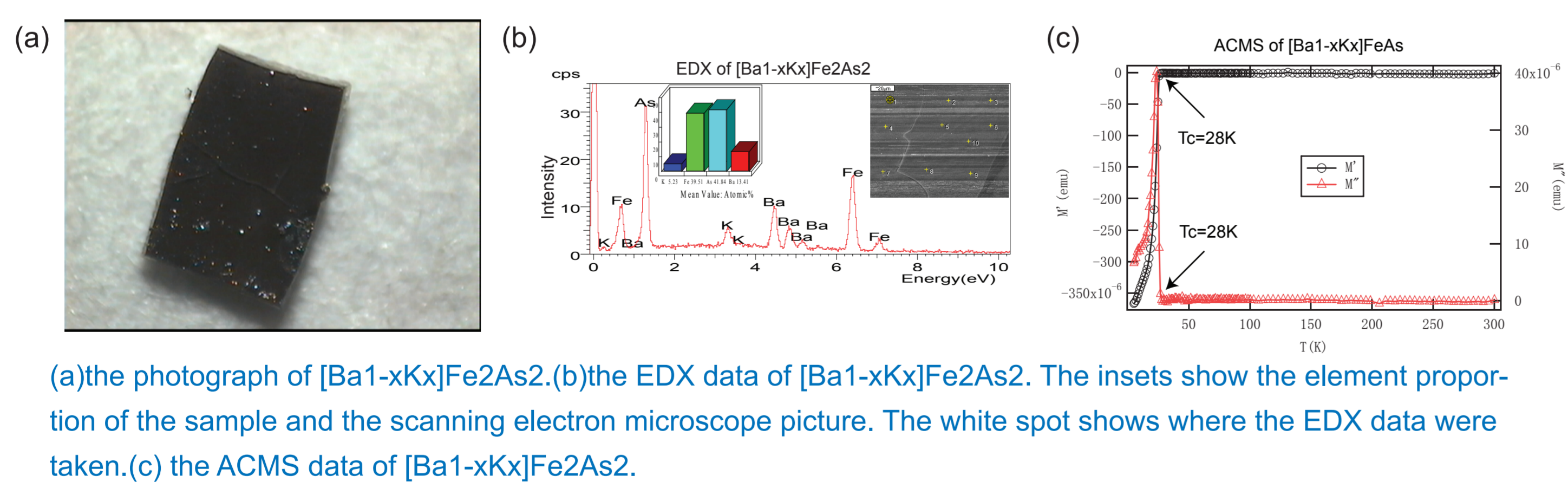


Temperature dependence of (a) the resistivity of polycrystal Nd_{0.8}Th_{0.2}OFeAs sample under external field and (b) Hall coefficient under an external field of 1 T.⁴

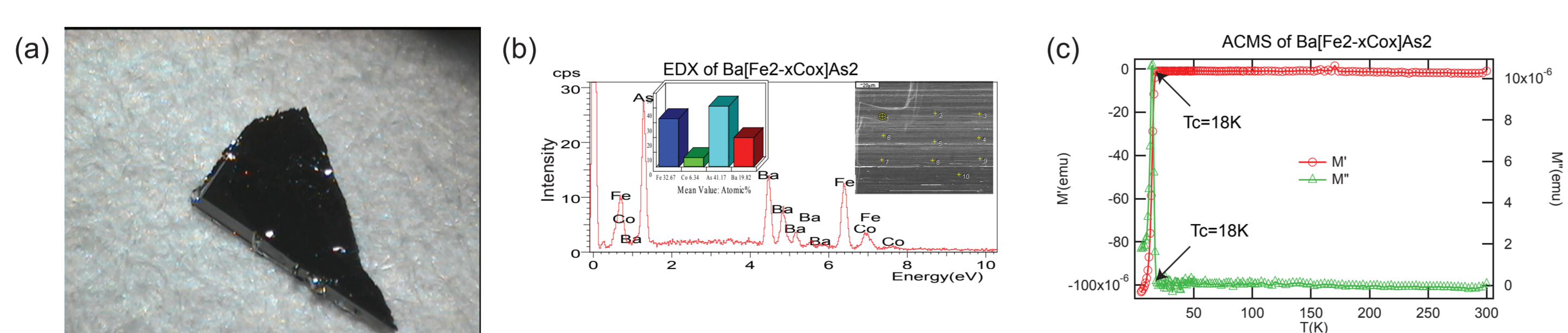
Ref 4. Min Xu, et al., Chemistry of Materials 01964h.R1 (2008).

the Single Crystals of the “122” System

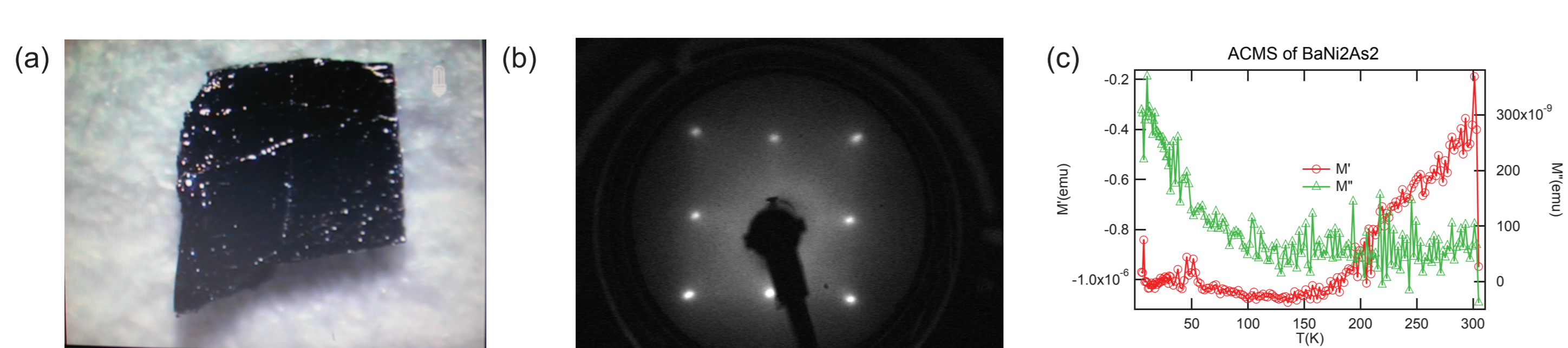
We use the Tin-flux and self-flux method to grow the single crystals of the “122” system. We mixed the starting material together according to their stoichiometric ratio (sometimes we put certain material excessively to make sure its doping level, for example, Potassium(K) in [Ba1-xKx]Fe2As2), then added Tin-flux or self-flux in proportional. The samples were annealed at high temperature and then slowly cooled down.



(a) the photograph of [Ba1-xKx]Fe₂As₂. (b) the EDX data of [Ba1-xKx]Fe₂As₂. The insets show the element proportion of the sample and the scanning electron microscope picture. The white spot shows where the EDX data were taken. (c) the ACMS data of [Ba1-xKx]Fe₂As₂.



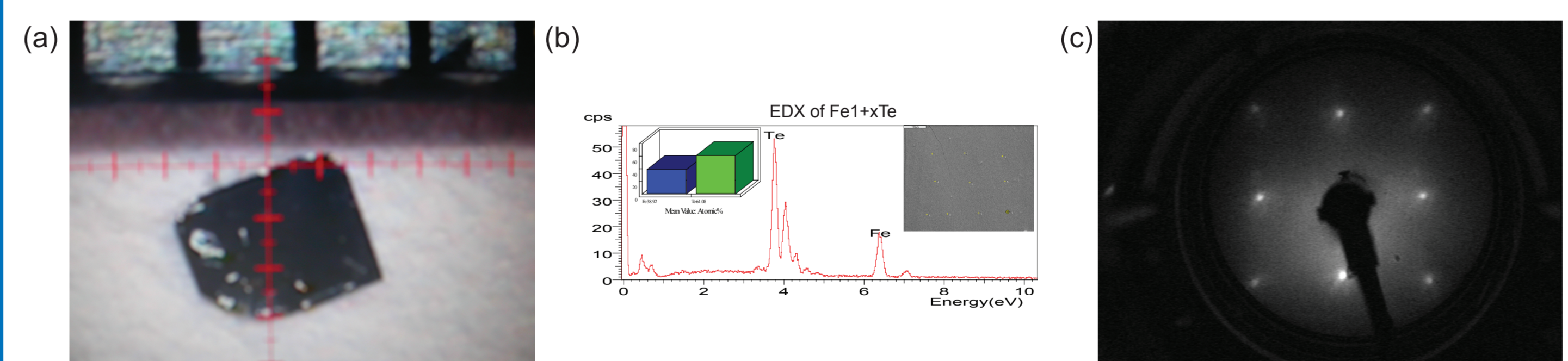
(a) the photograph of Ba[Fe_{2-x}Cox]As₂. (b) the EDX data of Ba[Fe_{2-x}Cox]As₂. The insets show the element proportion of the sample and the scanning electron microscope picture. The white spot shows where the EDX data were taken. (c) the ACMS data of Ba[Fe_{2-x}Cox]As₂.



(a) the photograph of BaNi₂As₂. (b) the LEED(low energy electron diffraction) pattern of BaNi₂As₂. (c) the ACMS data of BaNi₂As₂.

the Single Crystals of the “11” system

We use the NaCl-KCl-flux method to grow the single crystals of the “11” system. We firstly synthesized the polycrystal of Fe1+xTe(Se) by the solid state reaction method. Then we thoroughly mixed the polycrystal with NaCl, KCl together in proportional. Later the samples were annealed at high temperature and then slowly cooled down. At last, we could get the samples by washing them with deionized water.



(a) the photograph of Fe_{1+x}Te. (b) the EDX data of Fe_{1+x}Te. The insets show the element proportion of the sample and the scanning electron microscope picture. The white spot shows where the EDX data were taken. (c) the LEED pattern of Fe_{1+x}Te.

Summary

To summarize, we have made the polycrystals of the “1111” system, the single crystals of the “122” and “11” systems with different methods and tested their structures, compositions and transport properties. Since the new iron-based material has almost been discovered, people began to put more emphasis on the quality of the sample which mainly shows in its uniformity and purity. So our aim of the next period is to grow sample with high quality by all means.