

# Surface Ferromagnetism in HfO<sub>2</sub> Induced by Excess Oxygen Atoms

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## Abstract

First principles simulations based on density functional theory are performed to study surface magnetic properties of low index cubic, tetragonal, and monoclinic HfO<sub>2</sub> surfaces with different terminations. Our systematic calculations reveal that i) stoichiometric surfaces as well as Hf rich non-stoichiometric surfaces are non magnetic, and ii) O rich non-stoichiometric surfaces are ferromagnetic and often half metallic. The ferromagnetism found here is attributed to O surface electronic states with large O 2p spin exchange energy. Our finding provides a novel pathway to *d*<sup>0</sup> ferromagnetism for simple oxides with no magnetic ions involved. We further calculate the surface energy to discuss a possible reason for recent controversial observations of ferromagnetism in HfO<sub>2</sub>.

## Background

- (1) Temperature ferromagnetism is important both in its potential spintronics application and scientific research.
- (2) Lots of controversial of *d*<sup>0</sup> ferromagnetism in HfO<sub>2</sub>

## Simulation Method

Density functional theory calculations with plane wave bases

Code: VASP

$E_{xc}$ : PAW GGA-PBE

$E_{cut}$ : 500 eV

*k*-mesh: Monkhorst-Pack

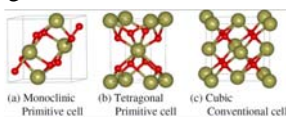
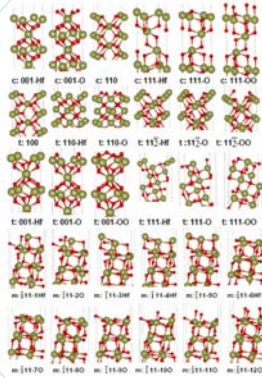
Surface: symmetrical slab

Spin polarized calculation

## Systems

Systems: low index cubic, tetragonal and monoclinic surfaces with different terminations

Halves of the symmetrical surfaces are given in the left figure.



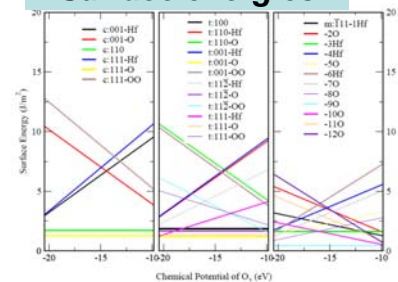
- (1) For Hf rich and stoichiometric surfaces, their ground state are non-magnetic.
- (2) Oxygen rich surfaces are magnetic.

For c:111-OO

$$\Delta E_{FM-AM} = -0.11 \text{ eV}$$

➤ The ground state is ferromagnetic.

## Surface energies

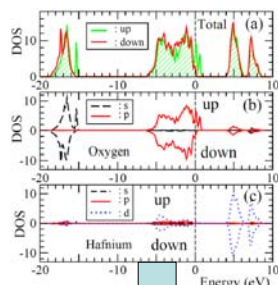
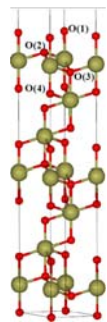


Stoichiometric m:-111-9O is the most stable. The non-stoichiometric surfaces could be stable under some chemical condition.

## Density of States

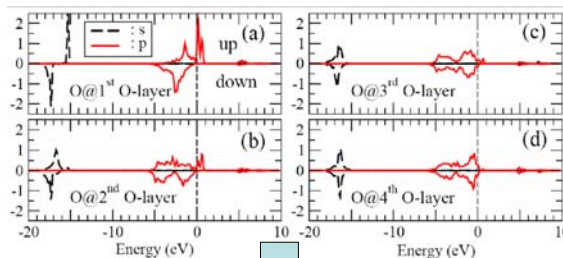
## layer by layer: spin-polarized

## non-spin-polarized

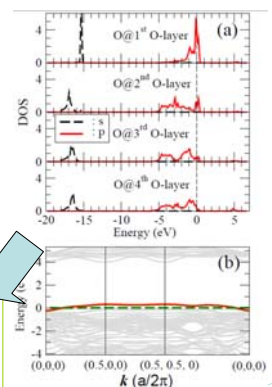


c:111-OO

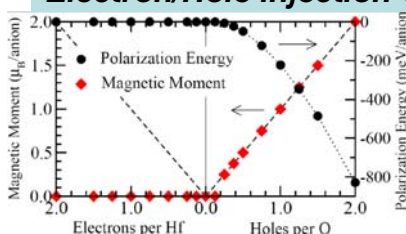
The surface is ferromagnetic, half metallic; Hole states are introduced near VBM, which are polarized O 2p orbitals.



The moment is localized on surface O atoms, which have sharp DOS peaks near Fermi level. => O 2p orbital large  $N(E_F)$ .



## Electron/Hole injection Spin exchange energy



Hole doping (O rich) is much easier to make HfO<sub>2</sub> polarized.

Spin exchange energy :  
Hf 5d orbital: ~1.04 eV  
O 2p orbital: ~3.03 eV  
Orbital radial distribution:  
O 2p < Hf 5d

$$\frac{1}{2}UN(E_F) > 1$$

➤ large U ➤ Stoner model of ferromagnetism

## Conclusion

- (1) The O rich surfaces could be stable under some chemical condition.
- (2) O rich surfaces can be ferromagnetic.
- (3) The ferromagnetism is attributed to the large surface O electronic states and large O 2p exchange energy.