Polar-Induced Electrostatic Control of Magnetism in

Multiferroics/Insulator Heterostructure: T-BiFeO₃/LaAlO₃ system Jie-Xiang Yu, Xin-Xin Zhao and Jing-Guang Che

We predict a strong electrostatic control of magnetic coupling in the heterostructure system with two insulating transition-metal oxides, of which one is multiferroics. By employing first principle calculation, we consider an example of the tetragonal-like BiFeO₃/LaAlO₃ (T-BFO/LAO) system to illustrate the magnetoelectric coupling of the first BiFeO₃ layer at interface. The total energy calculation with different magnetic states shows the different magnetic coupling with distinct interface terminal type and ferroelectric polarization. The study on the electronic structure shows gapless interface states and indicates an electronic reconstruction emerging at the each polarized interface, leading to the metallic $Fe^{(3-\delta)+}$ and $Fe^{(3+\delta)+}$ oxidation state respectively, which eventually exhibits different magnetic behaviors.







strong magnetoelectric coupling

Calculated in-plane and out-of-plane nearest-neighbor coupling parameters, J_{xv} and J_z , for bulk T-BFO and interfaces P1 and P2 in both I-La and I-Bi. Positive/negative values correspond to AFM/FM coupling, respectively. The local magnetic moments of Fe atoms have been normalized to S = 1.

(meV)	Bulk T-BFO	I-La		I-Bi	
		P1	P2	P1	P2
J_{xy}	65.3	-11.5	-30.7	60.5	-28.6



In structure I-La,(left)/I-Bi(right) (a) the total density of state (DOS) and orbital-resolved projected DOS (PDOS) for Fe atoms at the fourth FeO_2 layer from P1 (c), and at the interface P1 (b) and P2 (d). Positive/negative values correspond to the spin up/down channel. The t_{2g} and e_g characters are shown in different color. The Fermi energy is set at zero

Why are magnetic couplings different in I-La and I-Bi at interface P1?

The volume of octahedral BO_6 (B = Fe, Al) with various positions under (a) state S1 and (b) each ground magnetic state, S6 and S5, respectively, in both I-La and I-Bi structures. (c) Schematic diagram of Fe atoms at P1 and P2, where brown, green and violet balls correspond to Fe, La and Bi atoms, respectively. This shows that the absence



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